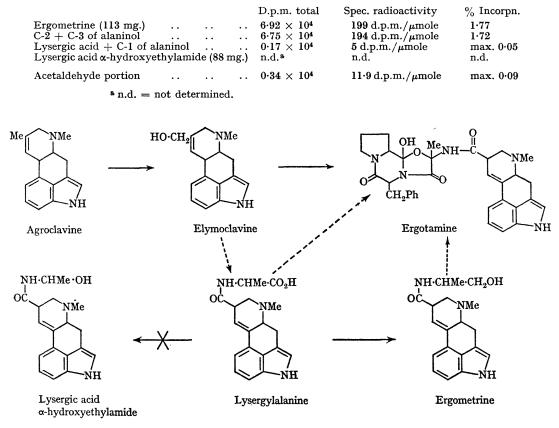
Biosynthesis of Ergot Alkaloids. Lysergylalanine as Precursor of Amide-type Alkaloids

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THE lysergic acid moiety of the amide-type ergot alkaloids arises from the structurally simpler clavines, *i.e.* from agroclavine *via* elymoclavine, although the detailed sequence of reaction steps leading from the latter to lysergic acid amide derivatives is not clear.¹ Based on findings that lysergic acid amide was not converted into the corresponding α -hydroxyethylamide by *Claviceps paspali*, and that fraction of the reaction mixture [silica gel G. (1) chloroform-ethanol, 9:1; (2) benzene-ethyl acetate-methanol, 7:2:1:], and hydrolysis of the D-lysergyl-L-alanine benzyl ester. The labelled material $(3.90 \times 10^6 \text{ d.p.m.}, 1.2 \,\mu\text{c}/\mu\text{mole})$ was added to two 100 ml. cultures of *Claviceps paspali.*³ After 4 days of fermentation, the two radioactive cultures and a nonradioactive parallel culture were

Incorporation of D-lysergyl-L-alanine labelled with ¹⁴C at the alanine 2-position into ergometrine and lysergic acid α-hydroxyethylamide by Claviceps paspali.



Biogenetic relationships among ergot alkaloids (full arrows: demonstrated; broken arrows: postulated).

L-alanine (but not ethylamine) was incorporated into the latter, Agurell^{1a} has suggested that lysergylalanine might be an intermediate in the formation of lysergic acid α -hydroxyethylamide and, because of the structural analogy, also of ergometrine and of the ergotamine-type peptide alkaloids. Further work by several groups is compatible with this suggestion.²⁻⁴

In the present study, D-lysergyl-L-alanine labelled with ¹⁴C at the 2-position of alanine was prepared by reaction of D-lysergic acid chloride hydrochloride with [2-¹⁴C]-DLalanine benzyl ester, t.l.c. resolution of the ether-soluble combined, and the alkaloids (201 mg. total) were isolated by extraction and crystallization as described earlier.³ The resulting crude crystalline alkaloid mixture had a specific radioactivity of 651 d.p.m./mg. and, according to t.l.c., contained an unusually high percentage (50–60%) of ergometrine in addition to the lysergic acid α -hydroxyethylamide. This agreed with the result of an enzymic determination⁵ of the amount of acetaldehyde released on treatment with buffer,³ which indicated the presence of 43–44% lysergic acid α -hydroxyethylamide. The acetaldehyde had only very little radioactivity (Table). A portion of the crude alkaloid was crystallized to constant specific radioactivity with carrier ergometrine and degraded by Kuhn-Roth oxidation to give acetic acid from C-2 and C-3 of the alaninol moiety.

The data (Table) indicate that lysergylalanine is incorporated into ergometrine but not into lysergic acid α -hydroxyethylamide. This incorporation is very specific in that extremely little radioactivity is scrambled into the ergoline portion but almost all of it is present in the expected position in the side-chain. This contrasts with the utilization of alanine in Claviceps paspali, which gives rise to substantial labelling of the ring system,^{2,3} and suggests that lysergylalanine is incorporated into ergometrine as a unit. This result favours one part of Agurell's suggestion, ^{1a} that lysergylalanine is an intermediate in the formation of ergometrine. The role of lysergylalanine in ergotamine formation has yet to be assessed. Its involvement in the formation of lysergic acid α -hydroxyethylamide, however, has become doubtful as a result of this experiment as well as of unpublished experiments by Agurell, who was also unable to demonstrate conversion of lysergylalanine (or its methyl ester) into the α -hydroxyethylamide.⁶

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¹ For recent review cf.: (a) S. Agurell, Acta Pharm. Suecica, 1966, 3, 71; (b) R. Voigt, Pharmazie (Berlin), 1968, 23, 285; 335; 419.